

Controlling pollution effects on metallic cultural heritage using linear sweep voltammetry

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AIMS

Linear Sweep Voltammetry (LSV) is a technique that allows the identification of the constituents of the thin tarnishing layers and its relative abundance [1, 2]. Minor amounts of the main compounds formed on the tarnish layer on silver and silver alloy samples exposed in the museum and in the chapel of Porto Cathedral, in Portugal, have been identified.

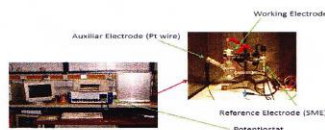
Tarnishing process and atmospheric pollutants

The tarnish process of silver and silver alloys is strongly dependent on the nature and levels of pollutants and on the climatic parameters such as: relative humidity (RH), average temperature, winds direction and intensity, etc.

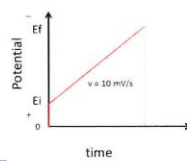
In spite of the amount of studies the tarnishing process of silver and silver alloys under natural atmospheres is not completely understood.

According to Graedel et al [3] hydrogen sulfide (H₂S) and carbonyl sulfide (COS) are the principal atmospheric aggressive gases for silver and copper leading to silver sulfide, however it is well documented in the literature that the formation of sulfides requires a minimum level of H₂S. Others such as Hallett et al [4] have identified the presence of several species other than silver sulfide.

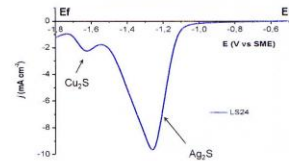
EXPERIMENTAL



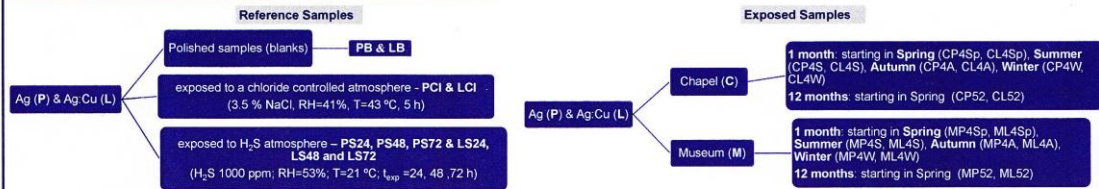
Linear Sweep Voltammetric Potential Profile



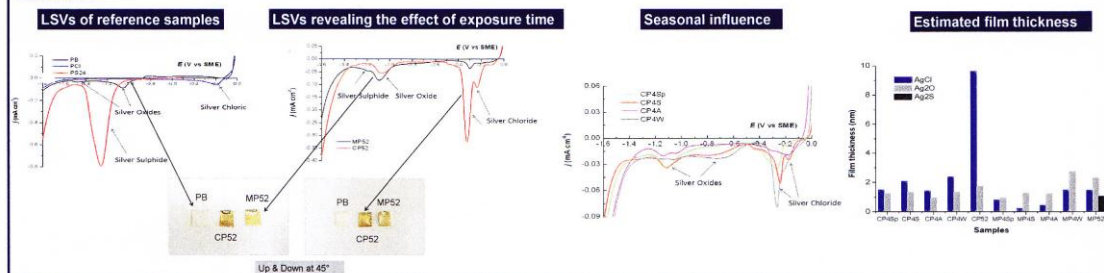
Cathodic Spectra



Working electrodes (made with exposed and non exposed samples)



RESULTS



CONCLUSIONS

LSV has identified as main components on all the samples as being silver chloride and silver oxides and only in a few samples silver sulfide and a mixed copper-silver sulfide. Film thicknesses ranging between 0.22 and 9.63 nm were estimated.

REFERENCES

- [1] Doménech-Carbó A, Doménech-Carbó MT, Costa V (2009) *Electrochemical methods applied to archaeometry, conservation and restoration* (Monographs in Electrochemistry Series, F. Scholz, ed) Springer, Berlin-Heidelberg, Berlin
- [2] Scholz F (2011) The electrochemistry of particles, droplets, and vesicles – the present situation and future tasks, *J Solid State Electrochem* 15:1699-1702
- [3] Graedel TE, Franey J P, Guillemer JG, Karmaliel GW, Mann DL (1985) On the mechanism of silver and copper sulfidation by atmospheric H₂S and COS, *Corros. Sci* 25:1163-1180.
- [4] Hallett K, Tickett D, McPhail DS, Chater RJ (2003) Application of SIMS to silver tarnish at the British Museum, *Appl Surf Sci* 203-204:789-792.

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