Mercury removal from aqueous solution and flue gas by adsorption on activated carbon fibres

Joa^o Valente Nabais a,*, P.J.M. Carrott a, M.M.L. Ribeiro Carrott a, Marisa Belchior a,

Dulce Boavida b, Tatiana Diall b, Ibrahim Gulyurtlu b

a Centro de Qui´mica de E´vora e Departamento de Qui´mica, Universidade de E´vora, Rua Roma`o Ramalho, 59, 7000-671 E´vora, Portugal b INETI, Estrada do Pac,o do Lumiar, 22, edif. J, 1649-038 Lisboa, Portugal Available online 15 December 2005

Abstract

The use of two activated carbon fibres, one laboratorial sample prepared from a commercial acrylic textile fibre and one commercial sample of

 $Kynol_1$, as prepared/received and modified by reaction with powdered sulfur and H_2S gas in order to increase the sulfur content were studied for

the removal of mercury from aqueous solution and from flue gases from a fluidized bed combustor. The sulfur introduced ranged from 1 to 6 wt.%

depending on the method used. The most important parameter for the mercury uptake is the type of sulfur introduced rather than the total amount

and it was found that the H₂S treatment of ACF leads to samples with the highest mercury uptake, despite the lower sulfur amount introduced. The

modified samples by both methods can remove HgCl₂ from aqueous solutions at pH 6 within the range 290–710 mg/g (ACF) which can be

favourably compared with other studies already published. The use of a filter made with an activated carbon fibre modified by powdered sulfur

totally removed the mercury species present in the flue gases produced by combustion of fossil fuel. # 2005 Elsevier B.V. All rights reserved.

PACS: 81.05.Rm; 81.05.zx; 82.80.Gk; 89.60.Ec

Keywords: Activated carbon fibres; Mercury adsorption; Sulfur impregnation; Surface chemistry