

Comprehensive Organic Chemistry Experiments for the Laboratory Classroom

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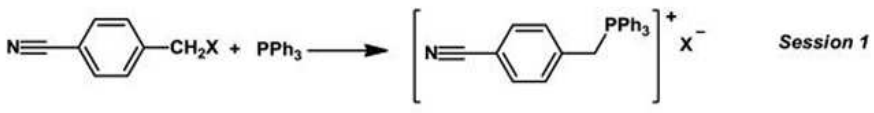
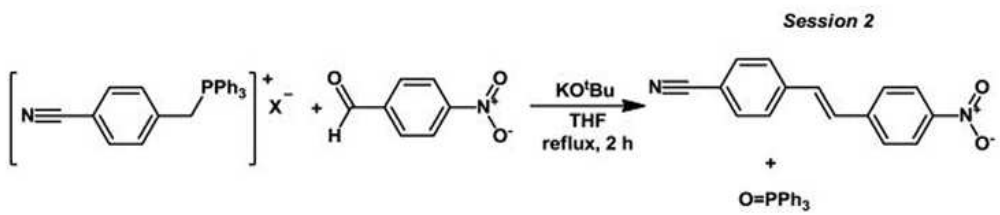


9.2.4. Building an Alkene Spacer by the Wittig Reaction: Synthesis of 4-[2-(4-Nitrophenyl)ethenyl]benzotrile

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 <p style="text-align: right;">Session 1</p>			
 <p style="text-align: right;">Session 2</p>			
Number of sessions (duration of each session)	Hazard level	Difficulty level	Level of study
2 (3 h)	Moderate	Medium	Intermediate
Class names Benzyl halides, phosphorus ylides, aldehydes, alkenes			
Concepts involved Wittig reaction, double bond formation, spectroscopic characterization, <i>E/Z</i> isomers, <i>E/Z</i> isomeric ratio			
Chemicals needed 4-Nitrobenzaldehyde, 4-(bromomethyl)benzotrile, triphenylphosphine, potassium <i>tert</i> -butoxide, toluene, THF, diethyl ether, dichloromethane, methanol, anhydrous magnesium sulfate			
Equipment and experimental techniques involved Heating mantle or oil bath with magnetic stirring, Büchner funnel, reflux apparatus, rotary evaporator, structural analysis (NMR, IR, MS)			
Keywords Aldehydes, alkenes, phosphorus ylides, precipitation, spectroscopic characterization, Wittig reaction			

Background

The Wittig reaction was discovered by George Wittig (Nobel Prize in Chemistry in 1979) in 1954.¹ The Wittig reaction is one of the most common and valuable methods for the formation of alkenes, starting from a carbonyl compound and a primary or secondary alkyl halide. The reaction involves the formation of an alkene from an aldehyde or a ketone using a phosphorous ylide derivative, $R_2C^--P^+(C_6H_5)_3$. This latter species is obtained from a nucleophilic substitution reaction of an alkyl halide by triphenylphosphine, followed by deprotonation of the