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"Losanga" decorated imitations of italic late republican black gloss tableware from South-Western Iberia: A multi-analytical/ microchemical characterization*

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ABSTRACT

The micro-chemical/mineralogical composition of samples of grey-paste imitations of Italic Late Republican black gloss tableware displaying a particular kind of lozenge-shaped decoration ("Losanga pottery") from Portuguese and Spanish archaeological sites in SW Iberia has been analysed by BSEM + EDS, μ XRD, Powder XRD, Portable XRF and µRaman spectroscopy. "Losanga" decorated ceramics have been found throughout the Western Mediterranean. Most of the sherds display a green-brown to greyish-black engobe at the surface resembling the gloss found in Attic pottery from Classical Greece. The overall chemical, mineralogical and fossiliferous homogeneities of the ceramic paste show common features (low K-feldspar/plagioclase ratio, high Ca content, abundance of well-preserved fragments of foraminifera microfossils) that indicate low firing conditions in the kiln ranging from 650 to 900 °C. With respect to the ceramic body, analytical results confirm an enrichment in the surface gloss layer of iron, potassium and aluminium and a depletion in silicon and calcium; the very fine grain size of the surface coating suggests elutriation of iron oxide-rich clays as confirmed by the presence of magnetite, maghemite and goethite in μ -XRD scan. Chemical and mineralogical data also suggest that the firing process was performed in a 600-850 °C temperature range, adopting the well-known technique of alternating oxidizing and reducing firing conditions largely employed at the time. The analytical results, while compatible with the archaeological hypothesis of a common provenance of the raw materials for pottery production from the Guadalquivir valley workshops cannot be considered conclusive due to the similarity in the geological substrate in the two SW Iberian regions under study.

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1. Introduction

In the study of Roman ceramics and pottery materials, microanalytical techniques combining chemical, microscopical and mineralogical analyses have proved very useful to shed light on raw material provenance and production technology [1–11]. Due to their widespread and ubiquitous distribution across the entire geographical width of the Roman influenced territories, amphorae fragments have received the most attention by researchers. In terms of socio-historical significance, though, tableware pottery production, albeit less studied, is of no less importance due to its wide distribution in the Mediterranean and because it may provide precious information on common trade routes and imported technologies across the Roman Empire [3,12–17].

With the term Italic black gloss ware (also known as Ceramica Campana, or Céramica Campaniense or "Ceramique Campanienne"

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according to its geographical distribution) archaeologists identify a specific group and type of tableware production developed between the IV and I century BC mainly in the Italic peninsula characterized by a red, creamy or greyish clay ceramic paste often associated with a black thin superficial engobe/gloss layer [3,12,15,17-19]. During the 2nd and 1st century BC, the massive commercialization of Italic black gloss tableware all around the Mediterranean basin led to imitative processes embedded with social and economic meanings in different regions of the Roman territories including SW Iberia [15,17]. These ceramics have been typologically regarded as tableware "imitations" of the Italic black gloss prototypes, supposed to be locally (in the sense of not imported from the Italic Peninsula) produced during the embryonic phase of interaction [15]. In fact, the craft imitative phenomenon during the Late Republican period in the provincial contexts is supposed to be the consequence of processes driven by the imports of exogenous goods stored in amphorae. Similarly, the growth of tableware imitation is due to the imports of Italic black gloss tableware during the last two centuries before the Common Era. Within the Italic black gloss production, a special typology of high socio-historical significance is represented by the so-called "Losanga-style" tableware,







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characterized by the presence of a decoration of stamped arcs joining four trefoils (Fig. 1). It is generally accepted that the "Losanga" decoration in pottery objects appeared in the 1st century BC [14] and had a major production centre in the Roman colony of Cales, in the northern part of modern Campania region in Italy [20,21]; "Losanga" pottery objects have been found throughout the Western Mediterranean including sites on the Atlantic African Coast [22]. In this study, samples of "Losanga" tableware from four archaeological sites in Southern/Central Portugal (Faro, Castro Marim, Monte Molião and Santarém) dated from the middle of the 1st century BC to the second half of the 1st century AD [17,19] and from 3 sites in the Seville area (Seville: Patio de Banderas; Italica: Pajar de Artillo; Orippo: Cortijo de Tixe) in the Andalusian region (Southern Spain) dated from the 1st century BC [12,16, 23] were selected for microchemical, microscopical and mineralogical analyses (Figs. 1, 2). A common feature of both Andalusian and Portuguese sites (which explains their role as important consumption centres of the time) lies in their proximity to the sea and to wellestablished ancient fluvial routes (respectively along the Guadalquivir, Tagus and Guadiana river valleys). The Guadalquivir area has long been recognized as an important centre for ceramic production [10,13, 23,24] as evidenced by the identification of several ceramic workshops with raw materials (clays and temper) for pottery manufacturing coming from the Quaternary fluvial sedimentary deposits of the Guadalquivir River. The importance of the Guadalquivir fluvial valley as a major ceramic production area in the Baetica province of the Hispania Peninsula during Roman occupation has been recently confirmed also by archaeometric analytical data on Haltern 70 amphorae [9]. It is worth noting that the current archaeological investigation in the Guadalquivir valley has not yet recognized the production of the tableware typology investigated in the current study. This state of affairs has led archaeologists to postulate a predominant unidirectional East to West flux of "Losanga" tableware produced in the Guadalquivir area from Baetic to Lusitania. In this vein, the assumption of a shared origin of the typologically similar "Losanga" tableware pottery fragments found in Portuguese sites with the Andalusian ones, though, has, so far, been based only on ceramological evidences [12] as well as on a sort of a "dependency" in terms of dynamism in the distribution of goods from the Andalusian region [25], following the attribution of a catalyst role to that region as expressed in the concept of the "Circulo del Estrecho" [26]. Major and trace-elements' chemical compositions of potsherds from Roman amphorae in the Sado and Tagus river valley [4,5] have indeed been compared with coeval pottery fragments from the Guadalquivir area: these studies identified similarities but also differences linked to changes in the geological/mineralogical terrain of the raw materials used in pottery manufacturing.



Fig. 1. "Losanga" decorated tableware pottery from SW Iberia.



Fig. 2. Sites location map.

The combined micro-analytical approach adopted in this study intends to add important data to the above mentioned debate by providing BSEM + EDS, μ XRD, PXRF and μ Raman spectroscopy data to compare the detailed chemical and mineralogical compositions of both the grey paste and the dark surface slip of selected samples from the investigated sites with the aim to establish for the first time a possible correlation between composition and typology, and between the Portuguese and Andalusian assemblages and to determine imported versus local manufacturing practices.

2. Materials and methods

31 samples of "Losanga style" tableware pottery (Fig. 1) were selected for analysis by BSEM + EDS, Portable XRF and Micro and Powder XRD. Resin inglobated polished blocks were analysed by BSEM-EDS using a HITACHI S3700N interfaced with a Quanta EDS microanalysis system. The Quanta system was equipped with a Bruker AXS XFlash® Silicon Drift Detector (129 eV Spectral Resolution at FWHM/MnK α). Standardless PB/ZAF quantitative elemental analysis was performed using the Bruker ESPRIT software. The operating conditions for EDS

Table 1

Concentration (wt.% for K₂O, CaO, Fe₂O₃ and ppm for Rb, Sr, Y, Zr and Nb) and standard deviation of the archaeological sample from Seville (SV), Italica (ITA), Orippo (OR), Castro Marim (CM), Santarém (ST), Monte Molião (MM), Faro (FA), and Castro Marim (CM).

Samples	K ₂ 0		CaO		Fe ₂ O ₃		Mn		Rb		Sr		Zr		Y		Nb	
	wt.%	St.dev	wt.%	St.dev	wt.%	St.dev	ppm	St.dev	ppm	St.dev	ppm	St.dev	ppm	St.dev	ppm	St.dev	ppm	St.dev
2 – SV	3.996	0.381	11.027	0.533	1.005	0.187	748.768	41.957	65.662	5.982	521.595	10.160	200.695	5.408	30.265	2.135	13.200	4.330
3 — SV	3.628	0.266	12.185	0.336	0.959	0.247	582.624	36.581	69.842	9.748	669.955	30.687	197.077	4.049	27.568	2.071	10.315	3.048
4 - SV	3.194	0.247	7.960	0.302	1.948	0.149	343.873	19.519	64.752	6.729	166.078	16.447	180.967	3.959	23.520	1.426	20.997	4.877
5 - SV	4.549	0.146	5.449	0.298	2.283	0.160	1130.729	46.999	87.731	9.706	719.726	20.857	187.300	2.411	82.373	11.179	4.237	2.420
6 – SV	5.159	0.126	5.200	0.235	1.863	0.171	863.487	22.805	69.808	6.863	570.299	35.024	207.224	4.529	28.393	3.853	11.698	2.721
7 – SV	3.098	0.113	11.165	0.315	0.918	0.294	595.597	51.398	63.038	9.329	454.909	27.059	212.681	3.417	32.929	0.830	11.517	3.445
8 – SV	6.247	0.220	7.907	0.445	1.905	0.101	719.755	90.330	89.100	4.492	968.304	40.169	192.281	2.920	65.458	3.493	n.d	n.d.
9 – SV	4.232	0.121	9.493	0.329	1.640	0.184	785.928	14.822	74.979	8.727	588.498	15.178	222.324	2.086	34.347	1.648	9.991	0.956
10 – SV	4.773	0.157	6.943	0.289	2.005	0.167	498.705	34.780	66.573	6.553	326.143	17.210	182.249	4.989	10.884	4.009	17.047	4.319
11 – SV	2.716	0.194	8.501	0.218	2.268	0.114	2097.862	52.875	64.042	3.374	270.616	16.689	202.083	1.317	27.611	0.619	15.215	2.136
12 – SV	4.344	0.273	6.853	0.480	2.352	0.065	434.203	68.340	101.021	4.015	322.423	15.914	169.154	1.249	22.723	0.524	14.366	1.083
13 — ITA	4.824	0.203	5.807	0.244	1.619	0.069	2576.995	59.597	72.940	4.900	982.622	27.309	198.067	3.467	33.917	1.663	4.031	3.995
14 – OR	4.290	0.176	9.065	0.239	1.220	0.085	1249.896	48.198	73.752	3.512	712.894	15.707	194.295	1.392	26.256	0.404	9.775	1.523
15 – OR	5.785	0.345	8.946	0.493	0.628	0.157	930.840	31.789	54.277	5.099	807.518	6.327	204.498	3.498	29.470	0.954	9.857	2.330
16 – OR	3.167	0.671	11.562	0.937	1.071	0.073	943.678	39.113	61.561	5.742	894.865	33.137	211.472	3.004	34.464	2.583	6.560	3.196
17 – OR	4.870	0.209	7.500	0.383	1.423	0.178	460.857	41.760	80.509	6.040	514.507	11.570	185.816	3.908	32.948	3.928	13.984	2.797
18 – OR	5.422	0.363	5.844	0.570	1.991	0.161	704.993	7.644	80.105	9.247	545.041	7.374	201.145	2.133	32.167	1.997	17.479	2.741
19 – CM	5.900	0.239	7.308	0.400	1.454	0.207	577.129	53.998	90.335	10.215	348.669	8.650	193.374	5.546	32.710	0.701	12.762	4.396
20 – ST	5.101	0.144	3.154	0.280	3.366	0.120	115.115	25.880	227.866	10.953	142.122	11.635	173.494	2.722	31.753	0.978	12.216	2.442
21 — ST	5.999	0.199	5.546	0.213	1.844	0.099	569.733	16.258	92.138	4.773	487.434	20.216	198.252	2.740	34.912	0.644	9.053	4.444
22 — ST	6.535	0.212	6.303	0.233	0.704	0.252	594.721	45.356	70.909	10.791	426.652	23.258	203.796	3.129	30.085	0.213	9.340	3.313
23 – ST	5.653	0.238	5.764	0.329	1.796	0.111	708.620	49.608	88.620	7.254	338.726	14.090	182.962	5.036	28.544	1.826	16.220	4.209
24 – ST	3.648	0.107	8.050	0.303	2.445	0.186	728.933	22.068	72.278	10.274	415.608	29.023	253.555	1.872	31.116	1.181	49.649	2.876
25 – ST	3.977	0.300	9.026	0.428	1.495	0.092	773.662	37.512	83.173	4.046	380.878	8.454	186.436	4.645	26.422	1.965	19.148	4.447
26 – MM	3.220	0.186	10.152	0.342	2.284	0.106	632.601	60.217	77.114	5.493	402.775	23.588	189.602	3.020	27.420	2.093	22.530	2.069
27 – MM	2.447	0.138	6.864	0.265	3.294	0.150	1646.575	14.106	132.799	10.280	370.455	16.989	181.306	2.544	41.805	1.189	19.855	2.667
28 – FA	7.228	0.165	5.765	0.130	1.789	0.093	957.865	21.185	73.880	4.126	561.278	12.931	200.301	4.346	32.485	1.577	11.124	3.225
29 – FA	6.284	0.357	3.382	0.579	2.398	0.126	1404.744	39.768	89.530	6.586	463.083	12.569	200.061	2.598	39.176	1.945	11.753	1.955
30 – CM	6.599	0.338	6.119	0.474	1.507	0.165	1017.040	46.066	79.448	6.322	527.135	21.295	191.826	1.811	31.868	4.766	14.512	3.004
31 – CM	3.934	0.164	7.918	0.359	2.276	0.126	1999.498	85.139	63.003	3.785	345.560	10.794	232.263	2.863	34.228	1.501	40.946	1.212
32 – CM	4.413	0.280	6.826	0.413	2.468	0.107	1245.596	59.200	68.308	4.518	414.771	8.894	205.039	4.071	30.549	0.725	18.912	3.887

analysis were as follows: BSEM mode (BSEM), 20 kV accelerating voltage, 10 mm working distance, and 120 μ A emission current. The detection limits for major elements (>Na) were in the order of 0.1 wt.%.

 μ -XRD and Powder XRD (PXRD) were used to identify crystalline phases. A Bruker AXS D8 Discovery XRD with the Da Vinci design with a Cu K α source operating at 40 kV and 40 mA and a lynxeye 1dimensional detector was used. Scans were run from 3 to 75° 20, with 0.05° step and 1 s measuring time by point. PXRD was used for bulk analysis while μ -XRD was used to investigate the black slip. The XRD system was reconfigured and a Goebel mirror and a 300 μ m beamcollimator were employed.

For major and minor elemental analyses a Portable XRF was used for quick and non-destructive examination. The instrument used was a Brüker TRACER III-SD handheld portable XRF spectrometer equipped with a Rhodium tube. Archaeological samples were analysed six times each with the same analytical condition. Six repetitions were considered sufficient, especially for ceramic samples, to obtain representative results for K, Ca, Fe, Mn, Rb, Sr, Zr, Y and Nb and to reduce problems related to grain size and mineralogy [7]. Flat (or almost flat) sections were selected for analysis to avoid interference caused by the presence of a slip, surface contaminant [8] and/or surface roughness [7]. Operating conditions were 40 kV and 35 µA using a filter (green filter, 12 mm Al + 1 mm Ti + 6 mm Cu with 300 s live-time-counts. Characteristic radiations were collected by a silicon drift detector (SDD) with an energy resolution of 149.68 eV at Mn K α FWHM. The beam spot is elliptical in shape with dimension 3 by 4 mm (7 mm²). The equipment also possesses a camera, allowing the visualization of the analysed area and thus the capture of the image and the spot of analysis. S1 PXRF software (v. 3.8.30) was used to record the spectra, and ARTAX software (v. 5.3.0.0) for the first spectra evaluation and for the selection of the reference standards.

Quantifications were performed converting count rates as the ratio of the K α Rh peak following the guideline of Speakman and Shackley [27] and using a "calibration macro" developed by Bruker, loaded in Excel environment. The calibration macro allows the selection of the spectra lines, in this case K K α , Ca K α , Mn K α , Fe K α , Rb K α , Sr K α , Y K α , Zr K α and Nb K α , and of inter-element corrections that compensate for the complex X-Ray physics occurring during excitation, notably matrix effects caused by inter element excitation and self absorption. To create the calibration 15 reference materials certified by the U.S. Geological Survey (USGS) were used; 7 were micro-analytical reference samples mounted in epoxy resin (BIR-1G, GSC-1G, GSE-1G, NKT-1G, BCR-2G, BHVO-2G, TB-1G) and 8 were powdered reference samples (BCR-2, W-2A, GSP-2, SBC-1, AGV-2, COO-1, OLO-1A, SGR-1B, BCR-2)



Fig. 3. Portable XRF data. Ternary diagram Rb-Y-Sr. All samples (but one from Santarem – ST20) show similar compositions.



Fig. 4. Portable XRF data. Binary diagram Rb/K₂O-Sr/CaO.

transformed into pressed pellets of 4 mm each. Previous studies on obsidian demonstrated that powdered and solid reference samples can be used for empirical calibration and data checking. To evaluate the accuracy of the calibration each of the 15 reference standard materials were measured six times and concentrations were calculated using the calibration. For each element the relative standard deviations (%RSDs) were plotted against certified values [28] showing that as concentration decreases, %RSD increases. For the chemical elements under consideration %RSDs obtained were higher than 20% for concentrations: $K_2O < 0.626$ wt.%; Mn < 176 ppm; Rb < 4.92 ppm; Sr < 32.3 ppm; Y < 4.8 ppm; Zr < 100 ppm; and Nb < 7.9 ppm. For CaO and Fe₂O₃ %RSD is less than 10%.

Micro-Raman spectroscopy was used to further complement BSEM + EDS and μ -XRD in investigating the surface black gloss. Two different modes of investigation were used: a) laser beam focused on powdered engobe material obtained by carefully scratching the surface of selected samples and b) laser beam focused directly on the black surface of the sherds. The RAMAN used was an HORIBA Xplora with an Olympus BX41 microscope; the laser was a He-Ne laser operating at 638 nm; two objectives of $10 \times$ and $100 \times$ were used to focus the laser beam on the samples. The backscattered light was dispersed by using a grating of 600 lines/mm with an exposure time of 5 s with 10 scans, and 1200 lines/mm with an exposure time of 10 s with 15 scans

Table 2 XRD Results.



Fig. 5. BSEM + EDS ceramic paste. General view of ceramic paste from Patio de Banderas site (Seville) showing bimodal grain size distribution in ceramic microfabric. Note abundance of calcareous microfossil fragments.

to get sufficiently informative spectra. The sample viewing system consisted of a colour television camera attached to the microscope. After each spectrum had been recorded, a visual inspection was performed in order to detect any superficial change caused by the laser. The HORIBA LabSpec Suite package was used for spectra acquisition and manipulation; the obtained spectra were identified through a comparison with the library of standard spectra of the software or with the reference spectra in the literature.

3. Results and discussion

3.1. X-ray fluorescence

Chemical data obtained by PXRF on 31 samples, 17 from Spain (11 from Patio de Banderas, 1 from Pajar de Artillo, 5 from Cortijo de Tixe) 14 from Portugal (6 from Santarém, 4 from Castro Marim, 2 from Faro and 2 from Monte Molião) are presented in Table 1. Notwithstanding the fact that the main elements present were known to be Si and Al coming from the alumino-silicate fraction of the clays and sands (quartz, phyllosilicates and feldspars: see XRD section below) used in ceramic manufacturing, for the purpose of comparing the composition of the potsherds from Spanish and Portuguese; selected elements for analysis were Ca, K, Fe (expressed in oxides %) and Mn, Rb, Sr, Y, Zr and Nb (expressed in ppm). Of the selected elements, CaO is the most abundant ranging from 3.18% (sample ST20 from Santarém) to 12.18%

	SEV2	SEV4	SEV5	SEV7	SEV10	SEV11	ITA12	CM18	ST19	ST20	ST21	ST23	MM26	FA29	CM32
Quartz	+++	+++	+++	+++	+++	+++	+++	+++	+++	+++	+++	+++	+++	+++	+++
Plagioclase	+/++	+/++	+/++	++	+/++	++	+/++	+/++	++	++	+/++	+/++	+/++	+/++	+/++
K-feldspar	+/++	+/++	+/++	+/++	+/++	+/++	++	+	+		+/++	+/++	+/++	+/++	+/++
Feldspathoids	_	_	_	_	_	_	_	_	_		_	_	_	_	_
Calcite	+++	+++	+++	+++	+++	+++	+++	++	++	+++	+++	+++	+++	+++	+++
Dolomite	++	++	vtg/?	vtg/?	++	++	vtg/?			+/++	vtg/?	+	++	++	vtg/?
Illite	++	++	++	++	++	++	++	++	++	++	++	++	++	++	++
Rutile								vtg	+						
Diopside	-	-	-	+	-	-	-				-	-	-	-	_
Analcime	-	-	-	-	-	-	-				-	-	-	-	_
Hematite	-	-	vtg	-	-	-	-	+			+/vtg	-	-	-	+/vtg
Smectite	_	_	_	_	+	_	_				_	_	+	_	_
Maghemite	-	-	-	-	-	-	-				-	-	-	+/vtg	+/vtg
Amphibole	_	_	?	_	_	_	_				_	_	_	_	_

+++: abundant, ++: present, + small amount, vtg: traces, ? doubts in presence, -: undetected.

Notes:

In sample 26, the smectite pattern is, in fact, the smectite-chlorite (PDF 07-0027) pattern.



Fig. 6. BSEM + EDS ceramic paste. General view of ceramic paste showing bimodal size in ceramic microfabric.

in sample SEV3 in Patio de Banderas site. According to the classification proposed by Maniatis & Tite [29] which sets 6% in weight of CaO as the percentage threshold to distinguish between calcareous and noncalcareous ceramics, 13 out of 17 Spanish samples and 9 out of 13 samples from Portugal can be considered as calcareous. Variations in CaO content may be ascribed to differences in the raw materials used (clays and sands), to intentional addition by the pottery workshops to produce objects with specific purposes [6,30,31] and/or to circulation of Ca-rich solutions within the pore spaces of the pottery fragments during their burial history [1]. The widespread presence (see BSEM + EDSsection below) in both Spanish and Portuguese potsherds under investigation of calcite (rarely dolomite) grains and of well preserved calcareous microfossils (foraminifera of the genus Globigerina) as main constituents of the temper fraction is supportive of the first hypothesis mentioned above. With a view to highlight compositional similarities between Spanish and Portuguese sites, a not so easy task bearing in mind the homogeneity of the used raw materials (estuarine river sediments) which in turn reflects the fairly homogeneous geological background across the Southern Portugal and Spain investigated sites, selected elemental binary and ternary plots have been used. Except for one sample from Santarém (ST20) and one from Monte Molião (MM26), both the ternary diagram Rb-Y-Sr (Fig. 3) and the Rb/K₂O-Sr/CaO binary diagram (Fig. 4) suggest a geochemical compatibility of the raw materials used in both Spanish and Portuguese samples. Sample



Fig. 8. BSEM + EDS ceramic paste. Planktonic foraminifera fragment with well preserved fossil microstructure from Patio de Banderas site (Seville).

ST20 in fact distinguishes itself for the high Rb content of 227 ppm versus an average value of 69 ppm in the Spanish samples.

3.2. X-ray diffractometry

XRD results from 16 samples (6 from Patio de Banderas, 1 from Italica, 4 from Santarém, 2 from Castro Marim, 1 from Monte Molião and 2 from Faro) are summarised in Table 2.

Mineralogically, XRD spectra are fairly homogeneous across Andalusian and Portuguese samples. Quartz is confirmed as the main crystalline phase present (abundant to very abundant) together with calcite, feldspars and phyllosilicates (mainly illites with minor illite/smectite and in one case, sample 10 from Patio de Banderas in Seville, smectite). Plagioclase feldspars are always more abundant than K-feldspars, this being especially true in Andalusian samples. Minor dolomite peaks are detectable in both Portuguese (Santarém, Monte Molião but not Faro and Castro Marim) and Patio de Banderas sherd fragments. With regards to kiln firing temperature, the near absence of newly formed "high temperature" mineral phases such as diopside, gehlenite and mullite (diopside has been found only in two samples -sample SEV 7 from Patio de Banderas and sample ST20 from Santarém, together with the presence of well defined peaks belonging to carbonates (calcite and dolomite) and the absence of melting features in phyllosilicates



Fig. 7. BSEM + EDS ceramic paste. Scattered micron-size monazite grains within ceramic microfabric from Santarem site.



Fig. 9. BSEM + EDS ceramic paste. Illitic laths splitting along cleavage planes due to dehydroxilation processes during firing processes. Hematite crystals also visible (bright).



Fig. 10. BSEM + EDS engobe. Interface engobe-ceramic paste.

typical of high temperature firing such as the development of mullite filled nanopores [10] suggests low firing conditions in the kilns (<900 °C). The widespread abundance of illite as the dominant clay mineral species is also in line with this hypothesis and may be used to fix the kiln temperature in the range >600 °C <950 °C [2,32]. Ti (rutile) and Fe oxides (hematite and maghemite) are present in all samples as accessory minerals as well as Cu-oxides (cuprite). The Fe-oxides may represent newly formed phases during the firing process and their coexistence is compatible with pottery manufacturing technique involving atmosphere changes during firing from oxidizing to reducing conditions [10].

3.3. Scanning-electron microscopy and micro-analysis

BSEM + EDS analysis has been used to describe the micro-texture of the ceramic paste and its relation with the black engobe surface layer and to identify the presence of individual mineralogical markers which due to their scarcity were undetectable by Portable XRF and XRD analyses.

3.3.1. Ceramic body

Under BSEM the ceramic paste in all samples displays a fine-grained microfabric with a bimodal grain-size distribution (Figs. 5,6). While the fine-grained groundmass is constituted by micron-size laths of illite and plagioclase feldspars, the main temper components are mineral grains of quartz and calcite with (in decreasing order of abundance)



Fig. 11. BSEM + EDS engobe. Micron-size Fe-rich inclusions.

titanite, zircon (often zoned), ilmenite, rutile, monazite, hematite and clinopyroxene mineral grains as accessory constituents. In Portuguese samples from Santarém (sample 19) and Castro Marim (sample 18) and in three Andalusian samples (sample 8 from Patio das Banderas in Seville city centre, sample 12 from Italica and sample 16 from Orippo), small inclusions of gold are also found in the temper fraction whereas in two samples from Seville (samples 1 and 7), scattered Pb-rich grains are present. The detection of Au grains in Roman pottery from the Guadalquivir river valley region is not unexpected as gold can be present in alluvial placer deposits used as a pottery raw material source [11]. Pyrite, (FeS₂) and chalcopyrite (CuFeS₂) grains were found only in samples from Santarém and Castro Marim. Common temper components, albeit found only as scattered micron-size mineral grains, are also grains of monazite ((Ce,La,Y,Th)PO4; Fig. 7), a phosphate accessory mineral typically found in fluvial deposits in areas characterized by granitic geological substrates such as is the case in the SW Iberian areas under investigation [33]. Microfossils (planktonic foraminifera of the genus Globigerina) are also commonly found with their microstructure still well preserved (Fig. 8): the widespread presence of calcite grains together with the preservation of the fossil outline and microstructure confirms XRD results in suggesting low firing temperatures (between 750 and 900 °C). The low firing temperature in the kiln used to produce the "Losanga" pottery under investigation is further confirmed by the absence of bright reaction rims formed by Ca-rich silicate phases (such as diopside, gehlenite and anorthitic plagioclases) which typically form in Ca-rich ceramics when fired at higher temperatures [2]. In terms of microfabric porosity in the temper, fissures at quartz grain boundaries as a result of shrinkage of clay minerals during drying and expansion of quartz crystals from α to β phases during firing [10,34] are present, particularly in samples from Andalucia where secondary porosity typically with rounded pores can also be seen. Illitic laths while maintaining their sheet-like typical habit, show separation along basal cleavage planes (Fig. 9), a feature that has been interpreted as a result of de-hydroxylation during low temperature firing processes [10].

3.3.2. Black gloss

A fairly continuous and homogeneous vitreous superficial engobe (thickness 13–16 µm) can be seen on both external and internal parts of all (but one) ceramic fragments (Fig. 10). With respect to the ceramic body, black gloss layers show Fe, K and Al enrichment and Si and Ca depletion with scattered micron-size Fe and Fe–Ti inclusions (Fig. 11). The fine grain size of the surface dark coating suggests the application by elutriation of iron oxide-rich clays rather than a varnish as confirmed by the presence of magnetite (Fe₃O₄), maghemite (γ -Fe₂O₃), and goethite (α -FeO(OH) in µ-XRD spectra which indicate highly reducing atmosphere and firing temperature not exceeding 850 °C). µ-Raman spectroscopy confirmed the presence of magnetite and hematite mineral species.

4. Conclusions

The chemical and mineralogical analyses of the tableware Losanga style fragments from Portuguese and Andalusian sites reveal the following common features: a) high Mg-calcite (rarely dolomite) content and high plagioclase/alkali feldspar ratio confirmed both by μ -XRD plots and by Portable XRF (high Sr and low Rb); b) abundance of well preserved planktonic (Globigerinae) foraminifera microfossil fragments; c) widespread presence of illitic as a mineral marker in the groundmass of the ceramic paste; and d) presence of minor mineral markers in the ceramic paste such as monazite (Ce,La,Y,Th)PO₄, pyrite FeS₂, chalcopyrite CuFeS₂, cuprite Cu₂O and gold.

The mineralogical data of the ceramic paste, in particular the presence of calcite grains, the good preservation of calcitic microfossils and the near absence of secondary newly formed minerals such as diopside, indicate a low technological process involving low firing temperature in the kiln (<900 °C). Small differences between Portuguese and

Andalusian ceramic fragments (rare presence in one Santarém sample of diopside indicating higher firing T) could be related to variable quality standards amongst local workshops.

The multidisciplinary approach adopted in this study which combined archaeological evidence with detailed microchemical, microscopical and mineralogical analyses provided a fast and nondestructive (or microdestructive) tool to add "hard" data towards answering a long lived and still very much alive debate in the archaeology of "Campaniense" pottery production in SW Iberia.

The fairly homogeneous nature of the geological substrate and of the raw materials available in the Portuguese and Spanish sites investigated makes it hard to draw unequivocal conclusions on a common local production of the potsherds in the Guadalquivir. Further research will focus on both expanding the number of investigated sites and refining trace element data using other techniques such as ICP-MS in order to better define trade routes, production technology transfer mechanisms in this important region of the Roman Empire.

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